Application of Pd^{II} -catalysed oxidative cyclisation of hydroxy(vinyl)furan: synthesis of (5*R*)-diastereomer of (+)-goniofufurone

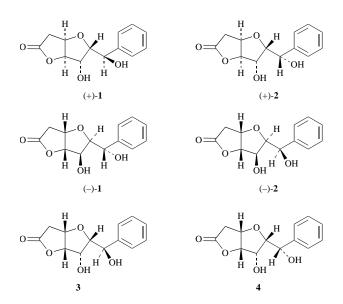
Hari Babu Mereyala,^{*,*a*} Rajendrakumar Reddy Gadikota^{*a*} and Ravikumar Krishnan^{*b*}

^a Organic Division III, Indian Institute of Chemical Technology, Hyderabad 500 007, India ^b Laboratory of Crystallography, Indian Institute of Chemical Technology, Hyderabad 500 007, India

A convenient synthesis of the (5R)-diastereomer of (+)-goniofufurone 5 from D-mannose has been completed. A key step is the intramolecular PdCl₂-catalysed oxidative cyclisation of hydroxy(vinyl)furan 6 to lactol 16.

Introduction

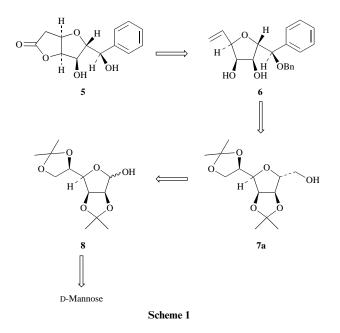
A novel styryl-lactone goniofufurone has been isolated ^{1,2} from ethanolic extracts of the stem bark of *Goniothalamus giganteus Hook. f.*, Thomas (Annonaceae) and shown to exhibit cytotoxic activity in tests with several human tumour cell lines.³⁻⁶ The constitution and relative configuration was assigned to the natural goniofufurone (+)-1 as D-glycero-D-ido, independently by Shing and Tsui⁷ and Jager and Gracza⁸ from the synthesis of both enantiomers. Thus, (+)-1, epi-goniofufurone (+)-2 and their enantiomers (-)-1 and (-)-2 have been synthesised.⁸⁻¹⁰ Enantiospecific synthesis of compounds 3 (3*S*,4*R*) and 4



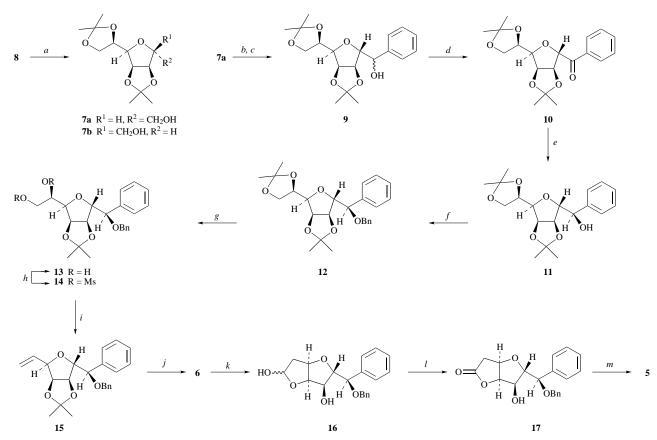
(3S,4R,7S) which are diastereomers of compounds (+)-1, (+)-2 has also been achieved.¹¹ Owing to their potential as antitumour agents several other related styryl-lactones have also been consequently isolated, characterised and their analogues synthesised.¹²

Results and discussion

Herein we report a convenient synthesis of yet another 5R diastereomer (compound **5**) of natural goniofufurone (+)-**1**. From retrosynthetic analysis (Scheme 1) compound **5** could be visualised as arising from hydroxy(vinyl)furan derivative **6** by means of the PdCl₂-mediated oxidative cyclisation protocol previously developed by us as a key step.¹³ Compound **6** in turn



could be realised from the C-glycoside 7a by involving oxidation of the primary alcohol followed by Grignard addition to obtain the desired C-7 stereocentre of target molecule 5. The vinyl group could be derived from the regioselective transformation of the primary acetonide by established methods. Compound 7a is obtainable from diacetone-D-mannose 8, involving C-C bond formation at C-1 to generate the required C-6 stereocentre of compound 5.14 C-2,3,4 stereocentres of D-mannose as such can be retained during these transformations to synthesise compound 5. The crystalline diacetone-D-mannose 8 was obtained from D-mannose in 85% yield¹⁵ (Scheme 2). Lactol 8 was treated with trimethylsulfoxonium iodide¹⁶ (TMSOI) and potassium tert-butoxide in dry dimethyl sulfoxide (DMSO) at room temp for 1 h to obtain a diastereomeric mixture of *C*-glycosides **7a**,**7b** in the ratio 4:1 (by 1H NMR spectroscopy) as a thick syrup in 79% yield. Fractional crystallisation of mixture 7a,7b (dichloromethane-hexane, 1:4) at 30 °C gave the required diastereomer **7a** as a crystalline compound (53%), mp 83–84 °C; $[a]_D - 12.6 \times 10^{-1} \text{ deg cm}^2 \text{ g}^{-1}$ (*c* 1.0, CHCl₃) [lit.,¹⁴ -10 (*c* 1.0, CDCl₃)]. Use of sodium hydride instead of KOBu^t resulted in the isolation of mixture 7a,7b in the ratio 2.3:1.14 Compound 7a was unambiguously characterised from ¹H NMR spectroscopy (400 MHz) and X-ray crystallography (Fig. 1) due to some minor differences observed while comparing the



Scheme 2 Reagents and conditions: a, TMSOI, KoBu', DMSO, room temp, 1 h; b, Oxalyl dichloride, DMSO, TEA, CH_2Cl_2 , -78 °C, 45 min; c, PhMgBr, THF, 0 °C, 3 h; d, PDC, CH_2Cl_2 , reflux, 3 h; e, NaBH₄, MeOH, -10 °C, 30 min; f, BnBr, NaH, DMF, 0 °C; 1 h; g, 60% aq. AcOH, room temp, 8 h; h, MsCl, TEA, CH_2Cl_2 , 0 °C, 30 min; i, NaI, butan-2-one, reflux, 8 h; j, 5% aq. H_2SO_4 , 1,4-dioxane, reflux, 2 h; k, PdCl₂(cat), CuCl, O₂, aq. DMF (1:4), room temp, 4 h; l, PDC, CH_2Cl_2 , reflux, 1 h; m, Pd–BaSO₄, H₂, MeOH, 6 h

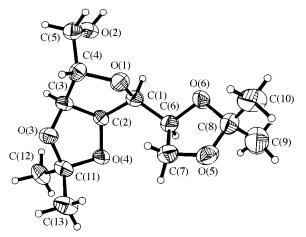


Fig. 1 An ORTEP drawing of compound **7a** with thermal ellipsoids (50% probability) and crystallographic numbering scheme

¹H NMR spectrum with that reported for **7a**.¹⁴ Swern oxidation of primary alcohol **7a** resulted in the formation of the corresponding aldehyde, which was immediately allowed to react in dry tetrahydrofuran (THF) with phenylmagnesium bromide at 0 °C for 3 h to obtain a diastereomeric mixture of compound **9**† as a syrup in 77% yield. Mixture **9** was characterised from the appearance of benzylic alcohol protons at $\delta \sim 4.7$ (1 H) and the aromatic protons (5 H) between δ 7.3–7.5. Oxidation of secondary alcohol **9** with pyridinium dichromate (PDC) in CH₂Cl₂ at reflux temperature for 3 h gave ketone **10** as a syrup, $[a]_{D}$ +47.4 (c 1.2, CHCl₃), in 79% yield. Formation of ketone 10 was evident from the appearance of a carbonyl absorption in the IR (neat) spectrum at 1655 cm⁻¹. The ¹H NMR spectrum of compound 10 indicated the absence of a benzylic alcohol proton and the appearance of a downfield singlet of δ 5.27 for the 2-H proton which is now α to the carbonyl group. Stereoselective reduction¹² of ketone 10 with sodium borohydride in methanol at -10 °C for 30 min gave the required diastereomer 11 as a thick syrup, $[a]_{D}$ +14.28 (c 1.4, CHCl₃), in 80% yield. Alcohol 11 was protected as its benzyl ether 12 by reaction with C₆H₅CH₂Br–NaH-dimethylformamide (DMF) at 0 °C for 1 h; ether 12 was obtained as a crystalline solid, mp 118–119 °C; $[a]_{D}$ +83.2 (c 1.0, CHCl₃), in 86% yield. Regioselective hydrolysis of the terminal isopropylidene group of compound 12 proceeded smoothly in aq. 60% acetic acid for 8 h at room temp to give the diol 13 as a syrup, $[a]_{D}$ +62.5 (c 1.2, CHCl₃), in 85% yield. Reaction of diol 13 with methanesulfonyl chloride (MsCl)triethylamine (TEA) in CH2Cl2 at 0 °C for 30 min gave the dimesyl derivative 14 as a thick syrup in 84% yield. Reductive elimination¹⁷ of diester 14 with sodium iodide in butan-2-one at reflux temperature for 8 h gave the vinylfuran derivative 15 as a syrup, $[a]_{D}$ +49.9 (c 1.4, CHCl₃), in 81% yield. Olefin 15 was characterised from the ¹H NMR spectrum from the appearance of vinylic protons (3 H) at δ 5.25 (1 H), δ 5.3 (1 H) and δ 5.8–6.0 (1 H). The isopropylidene group of compound 15 was deprotected by refluxing for 2 h in 1,4-dioxane containing aq. 5% H_2SO_4 to obtain the diol 6 as a syrup, $[a]_D + 44.8$ (c 0.5, CHCl₃), in 86% yield. Diol 6 was subjected to intramolecular oxidative cyclisation¹³ in water-DMF (1:4) containing PdCl₂ (0.2 mol equiv.), CuCl (1 mol equiv.) while oxygen was continuously bubbled through the solution for 4 h at room temperature to obtain, after work-up, a diastereomeric mixture of α/β lactols 16 in the ratio 1:2 (by ¹H NMR spectroscopy) as a thick syrup in 81% yield. Lactol 16 was carefully oxidised with PDC (1.0

[†] The diastereomeric mixture **9** was separated by flash chromatography to obtain compounds **9** (*threo*) and **11** (*erythro*) isomers. In the ¹H NMR spectrum of **9** (*threo*), H-7 appeared at δ 4.91 (t, $J_{6,7} = J_{7,OH} = 3.42$).

mol equiv.) in CH₂Cl₂ at reflux for 1 h to obtain the lactone 17 as a crystalline solid, mp 144–145 °C; $[a]_{\rm D}$ +133.3 (*c* 1.1, CHCl₃), in 76% yield. Lactone 17 was characterised from the appearance of a five-membered lactone absorption at 1784 cm⁻¹ in the IR(KBr) spectrum and also from the ¹H NMR (400 MHz) spectrum by the appearance of the 2-H axial proton at δ 2.68 (dd, J_{gem} = 18.8, $J_{2,3}$ = 1.9), the 2-H equatorial proton at δ 2.74 (dd, $J_{2',3}$ = 6.0) and the 4-H shifted downfield, due to the formation of a lactone, at δ 4.92 (dd, $J_{3,4}$ = 5.0, $J_{4,5}$ = 5.11). Finally the O-benzyl protecting group of compound 17 was deprotected by catalytic hydrogenolysis with 10% Pd–BaSO₄ in methanol under hydrogen for 6 h at room temp to give the 5*R*diastereomer of goniofufurone, compound **5** as a crystalline solid in 82% yield, mp 136–138 °C, $[a]_{\rm D}$ –32.4 (*c* 1.0, CHCl₃).

Experimental

¹H NMR spectra were measured with a Varian Gemini (200 and 400 MHz) spectrometer, with tetramethylsilane as internal standard for solutions in deuteriochloroform. *J*-Values are given in Hz. ¹³C NMR spectra were taken with a Varian Gemini (50 MHz) spectrometer with CDCl₃ as internal standard ($\delta_{\rm C}$ 77.0) for solutions in deuteriochloroform. Optical rotations were measured with a JASCO DIP-370 instrument, and [*a*]_Dvalues are in units of 10⁻¹ deg cm² g⁻¹. IR spectra were taken with a Perkin-Elmer 1310 spectrometer. Organic solutions were dried over anhydrous Na₂SO₄ and concentrated below 40 °C *in vacuo*.

3,6-Anhydro-1,2:4,5-di-*O*-isopropylidene-D-*glycero*-D-*manno*-heptitol 7a

A mixture of TMSOI (6.3 g, 28.8 mmol) and potassium tertbutoxide (2.8 g, 23.0 mmol) in dry DMSO (20 cm³) was stirred for 30 min at 10 °C. A solution of 2,3:5,6-di-O-isopropylidene-D-mannofuranose 8 (5.0 g, 18.2 mmol) in DMSO (10 cm³) was added to the above reaction mixture and the combined solution was brought to room temp and stirred for 30 min. When the reaction was complete, the mixture was quenched with saturated aq. ammonium chloride (75 cm³) and extracted into diethyl ether $(2 \times 100 \text{ cm}^3)$; the combined organic phases were washed with water $(2 \times 50 \text{ cm}^3)$, dried (Na_2SO_4) , and concentrated under reduced pressure to obtain a syrupy residue, which was filtered on a bed of silica gel [SiO2, 60-120 mesh; hexane-ethyl acetate (1:1)] to obtain a diastereomeric mixture of compounds 7a,7b (4.21 g, 79%) from which the *title compound* 7a (2.85 g, 53.4%) was separated by fractional crystallisation (dichloromethane-hexane; 1:4) as needles, mp 83-84 °C (Found: C, 56.84; H, 7.95. C₁₃H₂₂O₆ requires C, 56.92; H, 8.08%); [a]_D $-12.6 (c 1.0, \text{CHCl}_3); v_{\text{max}}(\text{KBr})/\text{cm}^{-1} 3600 (\text{OH}); \delta_{\text{H}}(400 \text{ MHz};$ CDCl₃) 1.35, 1.38, 1.45 and 1.51 (12 H, 4 s, 4 × CH₃), 3.61 (1 H, dd, $J_{7,7'}$ 11.4, $J_{6,7} = J_{6,7'} = 5.47$, 7-H), 3.64 (1 H, dd, 7-H'), 3.97 $(1 \text{ H}, \text{ dd}, J_{2,3}, 7.29, J_{3,4}, 3.9, 3-\text{H}), 4.08 (1 \text{ H}, \text{ dd}, J_{1,1'}, 9.6, J_{1,2}, 7.29, 3.9)$ 1-H), 4.09 (1 H, dd, J_{1',2} 7.26, 1-H'), 4.17 (1 H, dt, J_{5,6} 1.3, 6-H), 4.39 (1 H, ddd, 2-H), 4.67 (1 H, dd, J_{4.5} 6.0, 5-H) and 4.79 (1 H, dd, 4-H); δ_c(50 MHz; CDCl₃) 24.35, 24.85, 25.89 and 26.56 (4 × CH₃), 62.0, 66.40, 73.60, 80.93, 81.32, 82.42 and 84.78 (C-1 to -7) and 108.97 and 112.72 ($2 \times CMe_2$).

(7*R*/*S*)-3,6-Anhydro-1,2:4,5-di-*O*-isopropylidene-7-*C*-phenyl-Dglycero-D-manno-heptitol 9

Dry dichloromethane (25 cm³) containing DMSO (1.5 g, 14.2 mmol) was added to a solution of oxalyl dichloride (1.65 g, 13.1 mmol) in dry dichloromethane (10 cm³) at -78 °C and the resulting solution was stirred for 15 min. A solution of the alcohol **7a** (3.0 g, 10.9 mmol) in dry dichloromethane (5 cm³) was added to the above reaction mixture and the whole was stirred for 45 min, followed by the addition of TEA (3.3 g, 32.7 mmol). The reaction mixture was brought gradually to room temp and diluted with dichloromethane (50 cm³). The organic layer was washed with brine (2 × 25 cm³), dried (Na₂SO₄),

filtered, and evaporated to obtain the aldehyde as a yellow oil, and this was immediately treated with a solution of phenylmagnesium bromide (9.2 cm³, 14.3 mmol; 1.5 mmol dm⁻³) in dry THF (20 cm³) under nitrogen for 20 min. (Phenylmagnesium bromide was freshly prepared from bromobenzene and magnesium in THF.) The reaction mixture was then stirred for a further 3 h at 0 °C, quenched with saturated aq. ammonium chloride (25 cm³), and was filtered through Celite. The filtrate so obtained was concentrated to afford a syrupy residue and this was extracted into diethyl ether $(2 \times 75 \text{ cm}^3)$. The combined extracts were dried (Na₂SO₄), filtered, and concentrated to obtain an inseparable diastereomeric mixture of the title compound 9 (2.58 g, 77%) as a yellow syrup (Found: C, 65.07; H, 7.42. $C_{19}H_{26}O_6$ requires C, 65.12; H, 7.48%); $\delta_H(200$ MHz; CDCl₃) 1.25, 1.3, 1.32, 1.41, 1.42 and 1.45 (12 H, 6 s, $4 \times CH_3$), 3.95–4.2 (4 H, m, 1-H₂, 3- and 6-H), 4.25–4.4 (1 H, m, 2-H), 4.7-5.0 (3 H, m, 4-, 5- and 7-H) and 7.3-7.5 (5 H, m, ArH).

2,5-Anhydro-3,4:6,7-di-O-isopropylidene-1-C-phenyl-D-glycero-D-talo-heptose 10

PDC (5.37 g, 14.2 mmol) was added in one portion to a stirred solution of the diastereomeric mixture 9 (2.5 g, 7.1 mmol) in dry dichloromethane (50 cm³) containing powdered 4 Å molecular seives (0.3 g) at room temp. The reaction mixture was refluxed at 50 °C for 3 h. Progress of the reaction was monitored by TLC; when the reaction was complete, Celite (1.0 g) and diethyl ether (50 cm³) were added to the reaction mixture and the slurry was stirred at room temp for another 15 min, then was filtered through a bed of silica gel, and the solvent was evaporated off to obtain a crude residue, which was chromatographed [SiO₂, 60-120 mesh; hexane-ethyl acetate (4:1)] to obtain the title compound 10 (1.98 g, 79%) as an oil (Found: C, 65.41; H, 6.88. $C_{19}H_{24}O_6$ requires C, 65.50; H, 6.94%); $[a]_D$ +47.4 (*c* 1.2, CHCl₃); $v_{max}(neat)/cm^{-1}$ 1600 (C=O); $\delta_H(200)$ MHz; CDCl₃) 1.38, 1.4 and 2 × 1.42 (12 H, 3 s, 4 × CH₃), 3.85 (1 H, dd, J_{5,6} 7.4, J_{4,5} 4.5, 5-H), 3.9–4.15 (2 H, m, 7-H₂), 4.3–4.45 (1 H, m, 6-H), 4.78 (1 H, dd, J_{3,4} 5.3, 4-H), 5.27 (1 H, s, 2-H), 7.4–7.7 (3 H, m, ArH) and 8.01 (2 H, d, ArH); $\delta_{\rm C}$ (50 MHz; CDCl₃) 24.68, 25.13, 26.11 and 26.69 (4 × CH₃), 66.89, 73.04, 80.83, 2 × 82.68 and 85.21 (C-2/-7), 109.18 and 113.01 $(2 \times CMe_2)$, 128.63–134.13 (aromatic) and 196.20 (C=O).

(7*R*)-3,6-Anhydro-1,2:4,5-di-*O*-isopropylidene-7-*C*-phenyl-Dglycero-D-manno-heptitol 11

To a stirred solution of the ketone 10 (1.90 g, 5.45 mmol) in methanol (20 cm³) at -10 °C was added sodium borohydride (0.41 g, 10.9 mmol). The reaction mixture was quenched after 30 min with a few drops of acetic acid and was then gradually brought to room temp. Methanol was removed under reduced pressure and the resulting residue was extracted with diethyl ether $(2 \times 75 \text{ cm}^3)$. The combined extracts were dried (Na_2SO_4) , filtered, and concentrated to obtain a residue, which was chromatographed [SiO₂, 60-120 mesh; hexane-ethyl acetate (3:1)] to obtain the title compound 11 (1.5 g, 80%) as a syrup (Found: C, 65.01; H, 7.39. C₁₉H₂₆O₆ requires C, 65.12; H, 7.48%); [a]_D +14.3 (c 1.4, CHCl₃); $\delta_{\rm H}$ (400 MHz; CDCl₃) 1.29, 1.39 and $2 \times 1.47 (12 \text{ H}, 3 \text{ s}, 4 \times \text{CH}_3), 4.04 (1 \text{ H}, \text{dd}, J_{2,3} 7.45, J_{3,4} 3.7, 3-$ H), 4.06–4.13 (2 H, m, 1-H₂), 4.15 (1 H, dd, J_{5,6} 1.3, J_{6,7} 8.3, 6-H), 4.41 (1 H, ddd, $J_{1,2}$ 7.27, $J_{1',2}$ 7.25, 2-H), 4.56 (1 H, d, 7-H), 4.66 (1 H, dd, $J_{4,5}$ 6.0, 5-H), 4.83 (1 H, dd, 4-H) and 7.3–7.5 (5 H, m, ArH).

(7*R*)-3,6-Anhydro-7-*O*-benzyl-1,2:4,5-di-*O*-isopropylidene-7-*C*-phenyl-D-*glycero*-D-*manno*-heptitol 12

To a slurry of hexane-washed sodium hydride (0.15 g, 6.42 mmol) in dry DMF (5 cm^3) was added a solution of compound **11** (1.5 g, 4.28 mmol) in DMF (5 cm^3) at 0 °C. To this suspension was added dropwise benzyl bromide (0.87 g, 5.1 mmol) and the mixture was stirred for 1 h at room temp. When reaction

was complete, excess of sodium hydride was quenched by addition of methanol (1.0 cm³), and the reaction mixture was poured into ice-cold water (50 cm³) and extracted with diethyl ether $(2 \times 50 \text{ cm}^3)$. The combined extracts were dried (Na₂SO₄), filtered, and concentrated to obtain a residue, which was chromatographed [SiO₂, 60-120 mesh; hexane-ethyl acetate (4:1)] to yield the title compound 12 (1.66 g, 86%) as a crystalline solid, mp 118-119 °C (Found: C, 70.82; H, 7.27. C₂₆H₃₂O₆ requires C, 70.89; H, 7.32%); $[a]_{\rm D}$ +83.2 (c 1.0, CHCl₃); $\delta_{\rm H}(200 \text{ MHz};$ $CDCl_3$) 1.28, 1.32, 1.38 and 1.43 (12 H, 4 s, 4 × CH₃), 3.9–4.3 (6 H, m, 1-H₂, 2-, 3- and 6-H, and C₆H₅CHH), 4.39 (1 H, d, J_{4.5} 5.9, 5-H), 4.55 (1 H, d, J_{gem} 12.0, C₆H₅CHH), 4.7–4.75 (2 H, m, 4- and 7-H) and 7.25–7.48 (10 H, m, ArH); δ_c(50 MHz; CDCl₃) 24.80, 25.31, 26.20 and 26.92 (4 × CH₃), 67.05, 70.65, 73.72, 81.46, 82.06, 83.10 and 83.62 (C-1 to -6 and OCH₂Ph), 87.47 (C-7), 109.09 and 112.43 $(2 \times CMe_2)$ and 127.46–137.89 (aromatic).

(7*R*)-3,6-Anhydro-7-*O*-benzyl-4,5-*O*-isopropylidene-7-*C*-phenyl-D-glycero-D-manno-heptitol 13

A mixture of compound **12** (1.3 g, 2.95 mmol) and 60% aq. acetic acid (25 cm³) was stirred at room temp for 8 h. Reaction was monitored by TLC and, when complete, acetic acid was removed by azeotropic distillation with toluene *in vacuo* to obtain a crude, syrupy residue, which was chromatographed [SiO₂, 60–120 mesh; hexane–ethyl acetate (2:1)] to yield the *title compound* **13** (1.0 g, 85%) as a syrup (Found: C, 69.92; H, 6.99. C₂₃H₂₈O₆ requires C, 69.98; H, 7.05%); [*a*]_D +62.5 (*c* 1.2, CHCl₃); $\delta_{\rm H}$ (200 MHz; CDCl₃) 1.34 and 1.47 (6 H, 2 s, 2 × CH₃), 3.65–3.90 (3 H, m, 1-H₂ and 2-H), 4.02–4.25 (3 H, m, 3- and 6-H and C₆H₅CHH), 4.4 (1 H, d, *J*_{4,5} 5.9, 5-H), 4.55 (1 H, d, *J_{gem}* 12.0, C₆H₅CHH), 4.7–4.8 (2 H, d, 4- and 7-H) and 7.2–7.5 (10 H, m, ArH).

(7*R*)-3,6-Anhydro-7-*O*-benzyl-4,5-*O*-isopropylidene-1,2-bis-*O*-(methylsulfonyl)-7-*C*-phenyl-D-*glycero*-D-*manno*-heptitol 14

To a stirred solution of the diol **13** (0.9 g, 2.25 mmol) in dichloromethane (20 cm³) at 0 °C were added TEA (0.68 g, 6.7 mmol) and MsCl (0.61 g, 5.4 mmol) and the mixture was stirred for 30 min and brought to room temp. The reaction mixture was diluted with chilled water (20 cm³) and the organic layer was separated. The aqueous phase was extracted with dichloromethane (2 × 25 cm³) and the combined organic layers were dried (Na₂SO₄), filtered, and concentrated to obtain the *title compound* **14** (1.05 g, 84%) as a syrup (Found: C, 53.85; H, 5.69; S, 11.47. C₂₅H₃₂O₁₀S₂ requires C, 53.93; H, 5.79, S, 11.52%); $\delta_{\rm H}$ (200 MHz; CDCl₃) 1.30 and 1.45 (6 H, 2 s, 2 × CH₃), 2.98 and 3.05 (6 H, 2 s, 2 × SCH₃), 4.09–4.2 (2 H, m, 6-H and C₆H₅CHH), 4.39–4.45 (4 H, m, 1-H₂, 3- and 5-H), 4.55 (1 H, d, J_{gem} 12.0, C₆H₅CHH), 4.75–4.90 (3 H, m, 2-, 4- and 7-H) and 7.2–7.48 (10 H, m, ArH).

(7*R*)-3,6-Anhydro-7-*O*-benzyl-1,2-dideoxy-4,5-*O*-isopropylidene-7-*C*-phenyl-D-*altro*-hept-1-enitol 15

To a solution of dimesyl compound 14 (1.0 g, 1.79 mmol) in butan-2-one (25 cm³) was added sodium idodide (0.8 g, 5.4 mmol) and the reaction mixture was refluxed for 8 h. Progress of reaction was monitored by TLC; when the reaction was complete, solvent was removed under vacuum, saturated aq. sodium thiosulfate (20 cm³) was added, and the mixture was extracted with diethyl ether $(2 \times 50 \text{ cm}^3)$. The combined extracts were dried (Na₂SO₄), filtered, and concentrated under reduced pressure to obtain a syrupy residue, which was chromatographed [SiO₂, 60–120 mesh; hexane–ethyl acetate (5:1)] to obtain the title compound 15 (0.54 g, 81%) as an oil (Found: C, 75.27; H, 7.09. C₂₃H₂₆O₄ requires C, 75.38; H, 7.15%); [a]_D $+49.9 (c 1.4, CHCl_3); \delta_H(200 \text{ MHz}; CDCl_3) 1.3 \text{ and } 1.45 (6 \text{ H}, 2$ s, $2 \times CH_3$), 4.1-4.25 (2 H, m, 6-H and C₆H₅CHH), 4.39-4.48 (2 H, m, 3- and 5-H), 4.51-4.7 (3 H, m, 7-, 4-H and C₆H₅CHH), 5.25 (1 H, dd, J_{6,7trans} 10.0, 1-H), 5.32 (1 H, dd, J_{6,7cis} 16.5, 1-H'), 5.8-6.0 (1 H, m, 2-H) and 7.20-7.48 (10 H, m, ArH).

(7*R*)-3,6-Anhydro-7-*O*-benzyl-1,2-dideoxy-7-*C*-phenyl-D-*altro*-hept-1-enitol 6

To a solution of compound **15** (0.5 g, 1.36 mmol) in 1,4-dioxane (10 cm³) was added 5% aq. sulfuric acid (0.3 cm³) and the reaction mixture was refluxed for 2 h. After completion of the reaction the solvent was removed under vacuum and the residue obtained was extracted into ethyl acetate (50 cm³). The ethyl acetate phase was dried (Na₂SO₄), filtered, and concentrated under reduced pressure to obtain the *title compound* **6** (0.38 g, 86%) as a syrup (Found: C, 73.55; H, 6.72. C₂₀H₂₂O₄ requires C, 73.60; H, 6.79%); $[a]_D$ +44.8 (*c* 0.5, CHCl₃); $\delta_H(200 \text{ MHz}; \text{CDCl}_3)$ 3.95 (1 H, d, $J_{1,2}$ 3.5, 2-H), 4.09–4.22 (2 H, m, 3- and 4-H), 4.28–4.40 (2 H, m, 5-H and C₆H₅CHH), 4.52–4.67 (2 H, m, 1-H and C₆H₅CHH), 5.3 (1 H, dd, $J_{6,7trans}$ 10.0, 7-H), 5.35 (1 H, dd, $J_{6,7trans}$ 16.5, 7-H'), 5.82–6.02 (1 H, m, 6-H) and 7.22–7.50 (10 H, m, ArH).

(7*R*)-3,6-Anhydro-7-*O*-benzyl-2-deoxy-7-*C*-phenyl-D-*altro*-1,4-heptanolactol 16

To a solution of diol 6 (0.35 g, 1.07 mmol) in aq. DMF [10 cm^3 , DMF-water, 8:2] was added palladium(II) chloride (0.038 g, 0.2 mmol) and copper(I) chloride (0.1 g, 1.07 mmol). Oxygen was bubbled through the solution for 4 h at room temp. Progress of the reaction was monitored by TLC. When reaction was complete the reaction mixture was diluted with diethyl ether (75 cm³), filtered through a bed of silica gel, and eluted with diethyl ether. The extract was washed successively with 2% aq. hydrochloric acid (25 cm³) and finally with water (2 \times 25 cm³). The extract was dried (Na₂SO₄), and concentrated to obtain a residue, which was chromatographed [SiO₂, 60-120 mesh; hexane-ethyl acetate (4:6)] to obtain a diastereomeric mixture of the title compound 16 (0.29 g, 81%) as a thick syrup (Found: C, 70.03; H, 6.39. C₂₀H₂₂O₅ requires C, 70.16; H, 6.48%); δ_H(200 MHz; CDCl₃) 1.4–1.8 (1 H, br s, OH), 1.9–2.3 (2 H, m, 2-H₂), 3.7-4.35 (4 H, m, 3-, 5-, 6-H and C₆H₅CHH), 4.4-5.0 (3 H, m, 4-, 7-H, C₆H₅CHH), 5.5 (0.65 H, br d, 1-H^α), 5.63 $(0.35 \text{ H}, \text{t}, 1-\text{H}^{\beta})$ and 7.3–7.5 (10 H, m, ArH).

(7*R*)-3,6-Anhydro-7-*O*-benzyl-2-deoxy-7-*C*-phenyl-D-*altro*-1,4-heptanolactone 17

To a stirred solution of the lactol 16 (0.25 g, 0.73 mmol) in dry dichloromethane (25 cm³) containing powdered 4 Å molecular seives (0.05 g) was added PDC (0.28 g, 0.73 mmol) in one portion and the reaction mixture was refluxed for 1 h. After completion of the reaction, Celite (0.5 g) and diethyl ether (25 cm³) were added to the reaction mixture and the whole was filtered through a bed of silica gel and eluted with diethyl ether. The ethereal solution was concentrated to obtain a residue, which was chromatographed [SiO₂, 60-120 mesh; hexane-ethyl acetate (1:1)] to obtain the *title compound* **17** (0.19 g, 76%) as a crystalline solid, mp 144-145 °C (Found: C, 70.48; H, 5.85. $C_{20}H_{20}O_5$ requires C, 70.57; H, 5.92%); $[a]_D$ +133.0 (c 1.1, CHCl₃); v_{max} (KBr)/cm⁻¹ 1784 (C=O); δ_H (400 MHz; CDCl₃) 2.03 (1 H, d, J 7.6, OH), 2.68 (1 H, dd, J_{gem} 18.8, $J_{2,3}$ 1.92, 2-H), 2.74 (1 H, dd, $J_{2',3}$ 6.01, 2-H'), 4.01 (1 H, dd, $J_{5,6}$ 6.49, $J_{6,7}$ 3.37, 6-H), 4.29 (1 H, d, J_{gem} 12.0, C₆H₅CHH), 4.35 (1 H, ddd, J_{4,5} 5.11, 5-H), 4.52 (1 H, d, 7-H), 4.58 (1 H, d, C₆H₅CHH), 4.84 (1 H, ddd, J_{3,4} 5.0, 3-H), 4.92 (1 H, dd, 4-H) and 7.25-7.42 (10 H, m, ArH); $\delta_{c}(50 \text{ MHz}; \text{CDCl}_{3})$ 36.8 (C-2), 70.9, 73.18, 77.64, 80.8, 82.9 and 85.6 (C-3 to -7 and OCH₂Ph), 127.5–137.6 (aromatic) and 175.2 (C=O).

(7*R*)-3,6-Anhydro-2-deoxy-7-*C*-phenyl-D-*altro*-1,4-heptanolactone 5

Compound **17** (0.1 g, 0.29 mmol) was dissolved in methanol (20 cm³) to which 10% Pd–BaSO₄ (20 mg) was added and the mixture was stirred under hydrogen for 6 h. The catalyst was filtered off and the solvent was removed *in vacuo* to obtain the *title compound* **5** (0.06 g, 82%) as a crystalline solid, mp 136–138 °C (Found: C, 62.18; H, 5.61. $C_{13}H_{14}O_5$ requires C, 62.39; H,

5.64%); $[a]_{\rm D}$ – 32.4 (*c* 1.0, CHCl₃); $v_{\rm max}$ (KBr)/cm⁻¹ 1752 (C=O); $\delta_{\rm H}$ (400 MHz; CDCl₃) 2.08 (1 H, d, *J* 9.5, 5-OH), 2.42 (1 H, d, *J* 6.3, 7-OH), 2.72 (1 H, d, J_{gem} 17.8, 2-H), 2.82 (1 H, dd, $J_{2',3}$ 6.1, 2'-H), 3.98 (1 H, dd, $J_{5,6}$ 8.0, $J_{6,7}$ 3.37, 6-H), 4.39 (1 H, ddd, $J_{4,5}$ 4.61, 5-H), 4.82 (1 H, dd, 7-H), 4.9 (1 H, dd, $J_{3,4}$ 4.61, 3-H), 4.99 (1 H, t, $J_{3,4} = J_{4,5}$ 4.61, 4-H), 7.25–7.40 (5 H, m, ArH); $\delta_{\rm C}$ (50 MHz, CDCl₃) 36.96 (C-2), 71.8, 72.69, 76.54 (merged), 83.12, 84.15 (C-3 to -7), 127.5–137.6 (aromatic), 175.6 (C=O).

X-Ray structure determination

Crystal data of compound 7a. C13H22O6, needles, orthorhombic. Space group $P2_12_12_1$, a = 5.392(1), b = 10.757(2), c = 24.716(4) Å, V = 1433.6(4) Å³, Z = 4, $D_c = 1.271$ g cm⁻³, T = 293 K, crystal dimensions $0.12 \times 0.12 \times 0.17$ mm, 1185 reflections measured, $2\theta_{\text{max}} = 45^{\circ}$, $0 \le h \le 5$, $0 \le k \le 11$, $0 \le l \le 26$, 1157 unique reflections, and 826 with $I \ge 3\sigma(I)$, $\mu = 0.10 \text{ mm}^{-1}$. Final R = 0.063 (172 parameters and unit weights), maximum shift/error = 0.001, $\Delta \rho_{\text{max}} = 0.14$ e Å⁻³, $\Delta \rho_{\text{min}} = -0.12$ e Å⁻³. Mo-K α radiation ($\lambda = 0.710$ 73 Å), graphite crystal monochromator, Siemens R3m/V four-circle diffractometer (ω -2 θ scan technique). Data corrected for Lorentz and polarization effects but not for absorption. The structure was solved by direct methods (SHELXTL-Plus)¹⁸ and non-H-atoms were anisotropically refined. H-Atoms were included at geometrically calculated positions and refined as riding atoms with fixed isotropic temperature factors (U = 0.08Å²). Fig. 1 was prepared using SHELXTL-Plus. All calculations were made on a MicroVax-3100 workstation. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre (CCDC). See Instruction for Authors, J. Chem. Soc., Perkin Trans. 1, 1997, Issue 1. Any request to the CCDC for this material should quote the full literature citation and the reference number 207/154.

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References

- 1 X. P. Fang, J. E. Anderson, C. J. Chang, P. E. Fanwick and J. L. McLaughlin, J. Chem. Soc., Perkin Trans. 1, 1990, 1655.
- 2 X. P. Fang, J. E. Anderson, C. J. Chang, J. L. McLaughlin and P. E. Fanwick, J. Nat. Prod., 1991, 54, 1034.
- 3 A. A. E. Eizayat, N. R. Ferrigni, T. G. McCloud, A. T. McKenzie, S. R. Byrn, J. M. Cassady, C. J. Chang and J. L. McLaughlin, *Tetrahedron Lett.*, 1985, 26, 955; A. Alkofahi, W. W. Ma, A. T.

McKenzie, S. R. Byrn and J. L. McLaughlin, J. Nat. Prod., 1989, **52**, 1371; X. P. Fang, J. E. Anderson, C. J. Chang and J. L. McLaughlin, *Tetrahedron*, 1991, **47**, 9751; J. L. McLaughlin, C. J. Chang and D. L. Smith, *Stud. Nat. Prod. Chem.*, 1991, **9**, 383; in *Human Medicinal Agents From Plants*, ed. A. D. Kinghorn and M. E. Balandrin, American Chemical Society, Washington, DC, 1993, p.112.

- 4 X. P. Fang, J. E. Anderson, X. X. Qui, J. F. Kozlowski, J. F. Chang and J. L. McLaughlin, *Tetrahedron*, 1993, 49, 1563.
- J. R. Hlubucek and A. V. Robertson, Aust. J. Chem., 1967, 20, 2199;
 H. Achenbach and G. Wittmann, Tetrahedron Lett., 1970, 3259;
 K. Jewers, J. B. Davis, J. Dougan, A. H. Manchada, G. Blunden, Aye Kyi and S. Wetchpinan, Phytochemistry, 1972, 11, 2025;
 T. W. Loder and R. H. Nearn, Heterocycles, 1977, 7, 113.
- 6 S. K. Talapatra, D. Basu, T. Deb, S. Goswami and B. Talapatra, *Indian J. Chem., Sect. B*, 1985, 24, 29; T. W. Sam, C. Seu-Yeu, S. Matsjeh, E. K. Gan, D. Razak and A. Mohamed, *Tetrahedron Lett.*, 1987, 28, 2541.
- 7 T. K. M. Shing and H. C. Tsui, J. Chem. Soc., Chem. Commun., 1992, 432.
- 8 T. Gracza and V. Jager, Synlett, 1992, 191.
- 9 T. K. M. Shing, H. C. Tsui and Z. H. Zhou, *Tetrahedron*, 1992, 48, 8659.
- 10 T. Gracza and V. Jager, Synthesis, 1994, 1359.
- 11 T. K. M. Shing and H. C. Tsui, *Tetrahedron: Asymmetry*, 1994, 5, 1269.
- 12 T. K. M. Shing, H. C. Tsui and Z. H. Zhou, J. Chem. Soc., Chem. Commun., 1992, 810; P. J. Murphy, J. Chem. Soc., Chem. Commun., 1992, 1096; C. Mukai, I. J. Kim and M. Hanaoka, Tetrahedron Lett., 1993, 34, 6081; M. Tsubuki, K. Kanai and T. Honda, Synlett., 1993, 653; K. R. C. Prakash and S. P. Rao, Tetrahedron, 1993, 49, 1505; J. Ye, R. K. Bhatt and J. R. Falck, Tetrahedron Lett., 1993, 34, 8007; P. J. Murphy and S. T. Dennison, Tetrahedron, 1993, 49, 6695; Z. C. Yang and W. S. Zhou, J. Chem. Soc., Perkin Trans. 1, 1994, 3231; Tetrahedron, 1995, 51, 1429; T. K. M. Shing, H. C. Tsui and Z. H. Zhou, J. Org. Chem., 1995, 60, 3121; D. Xu and K. B. Sharpless, Tetrahedron Lett., 1994, 35, 4685; T. K. M. Shing, V. W. F. Tai and H. C. Tsui, J. Chem. Soc., Perkin Trans. 1, 1994, 1293; S. Y. Ko and J. Lerpiniere, Tetrahedron Lett., 1995, 36, 2101; J. P. Surivet and J. M. Vatele, Tetrahedron Lett., 1996, 37, 4373; C. Mukai, I. J. Kim, M. Kido and M. Hanaoka, Tetrahedron, 1996, 52, 6547; Z. C. Yang and W. S. Zhou, Heterocycles, 1997, 45, 367.
- 13 K. Krishnudu, P. R. Krishna and H. B. Mereyala, *Tetrahedron Lett.*, 1996, 37, 6007.
- 14 C. Frechou, L. Dheilly, D. Beaupere, R. Uzanet and G. Demailly, *Tetrahedron Lett.*, 1992, 33, 5061.
- 15 O. T. Schmidt, Methods Carbohydr. Chem., 1963, 2, 319.
- 16 E. J. Korey and M. Chaykovsky, J. Am. Chem. Soc., 1965, 87, 1353.
- 17 J. K. N. Jones and J. L. Thomson, Can. J. Chem., 1957, 35, 955.
- 18 G. M. Sheldrick (1990). SHELXTL-Plus, Revision 4.11/v Siemens analytical X-ray Instr., Inc., Application Lab., Single Crystal, Siemens AG, AUT V353, Karlsruhe, Germany.

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